

Off-Stoichiometry-Induced Surface Protective Layer for High Performance Layered Lithium Cobalt Oxide

Hyeon Jeong Lee^a, Jang Wook Choi^a

^a *School of Chemical and Biological Engineering and Institute of Chemical Processes, Seoul National University, Republic of Korea*

E-mail: hyeonjeongyi37@gmail.com

Despite the long history in commercial lithium-ion batteries, layered lithium cobalt oxide (LiCoO₂) suffers from structural degradations that shorten the cycle life when operating at high voltages (i.e., 4.5 V vs Li/Li⁺) in organic electrolytes or even at moderate voltages in aqueous electrolytes. This limited performance originates from the O3-to-O1 phase transition involving cobalt ion dissolution or surface oxidation followed by Li₂O leaching, respectively. Here, we report a one-pot synthesis that yields LiCoO₂ bearing a spinel-Co₃O₄ surface structure with a thickness of 2 nm *via* lithium-deficient stoichiometry (Li:Co=0.98:1). The lithium-deficiency induces the spinel structure, a thermodynamically preferred phase at the given stoichiometry, by temperature-specific phase separation. The cobalt ion in the spinel layer is at the oxidation state of 2+ or 3+ and thus mitigates its dissolution and oxygen gas evolution even at fully charged state where the oxidation state of cobalt in the bulk reaches 4+, improving cyclability in both organic and aqueous electrolytes markedly.

References:

- [1] K. Mizushima, P. Jones, P. Wiseman, J. B. Goodenough, *Mater. Res. Bull.* *15*, (1980) 783-789.
- [2] Winter, J. O. Besenhard, M. E. Spahr, P. Novák, *Adv. Mater.* *10*, (1998) 725-763.
- [3] M. S. Whittingham, *Chem. Rev.* *104*, (2004) 4271-4302.