

# High voltage Li-ion cell development based on aqueous processing of mixed manganese phosphate cathode and stable electrolytes

Iratxe de Meatzza<sup>a</sup>, Aitor Eguia-Barrio<sup>a</sup>, Idoia Urdampilleta<sup>a</sup>, Tong Zhang<sup>b</sup>, Niloofar Ehteshami<sup>b</sup>, Elie Paillard<sup>b</sup>, Jean-Baptiste Ducros<sup>c</sup>, Willy Porcher<sup>c</sup>, Diogo Vieira Carvalho<sup>d</sup>, Varvara Sharova<sup>d</sup>, Arianna Moretti<sup>d</sup>, Stefano Passerini<sup>d</sup>

<sup>a</sup> CIDETEC, Parque Tecnológico de San Sebastián, Paseo Miramon 196, 20014, Donostia-San Sebastian, Spain

<sup>b</sup> Helmholtz Institute Münster – Forschungszentrum Jülich (IEK-12), Correnstrasse 80, 48149 Münster, Germany

<sup>c</sup> CEA-LITEN, 17 avenue des Martyrs, 38000 GRENOBLE, France

<sup>d</sup> Helmholtz Institute Ulm (HIU), Helmholtzstrasse 11, 89081 Ulm, Germany

E-mail: imeatza@cidetec.es

The development of highly efficient batteries is a critical need in automotive industry in order to enable the future success of Electric Vehicles (EV). Active material with higher insertion potential in association with a compliant electrolyte play a very important role for increasing Li-ion cell energy density and, as a consequence, EV autonomy. Moreover, green process like water-based slurries for positive electrode manufacturing allows significant reduction of environmental impact and cost. This work, enclosed in the European SPICY project, presents results for mixed manganese and iron phosphate,  $\text{LiFe}_{1-x}\text{Mn}_x\text{PO}_4$  (LFMP), polyanionic cathode with waterborne binders and electrolyte systems stable at high voltage, aiming to develop safe cells with 20% increase in energy density and 20% cost reduction from the  $\text{LiFePO}_4$  (LFP) baseline.

Polymer binders classically used on the negative electrode, such as Styrene Butadiene Rubber (SBR), are not electrochemically stable at the cathode operating voltages (starts to decompose at 4.2 – 4.3 V). Thus the electrochemical stability window of several waterborne polymeric binders has been analyzed by cyclic voltammetry in 3-electrode Swagelok cells: an acrylate-based latex by Zeon® seems to be stable up to 5 V vs  $\text{Li}^+/\text{Li}$ . Several LFMP active materials with Mn content in the range  $0.3 < x < 0.7$  have been thus formulated in aqueous slurries with sodium carboxymethyl cellulose (CMC) and acrylate-based latex binders. Positive electrodes based on LFP and LFMP (89 wt.% active mass) have been prepared showing suitable mechanical properties for high areal loadings up to  $2.4 \text{ mAh cm}^{-2}$ . Graphite electrodes of 95 wt.% active mass loading and  $2.65 \text{ mAh cm}^{-2}$  (water-based CMC/SBR formulation) have been used in full coin cells to assess electrochemical performance with different additives (VC, FEC, LiTFSI), carbonates and new solvent-based (sulfolane, SL and adiponitrile, ADN) electrolytes. Best performing cathode has been achieved for the active material composition  $\text{LiFe}_{0.45}\text{Mn}_{0.55}\text{PO}_4$  ( $155 \text{ mAh g}^{-1}$ ;  $>600 \text{ Wh kg}^{-1}$ ). High initial coulombic efficiencies are obtained (83-85%) in full Li-ion cells for several of the studied electrolytes, allowing for  $140 \text{ mAh g}_{(\text{LFMP})}^{-1}$  at low discharge-rates (C/20) and ca.  $115 \text{ mAh g}_{(\text{LFMP})}^{-1}$  at high discharge rate (3C). Cycle-life tests at 1C charge/discharge (100% DoD, voltage window 2.5 – 4.4 V) have shown stable cycling behavior with a delivered capacity of  $\sim 120 \text{ mAh g}_{(\text{LFMP})}^{-1}$  and capacity retention above 90% after 300 cycles for three electrolytes: 1M  $\text{LiPF}_6$  EC:DMC:PC +2wt%VC, 1M  $\text{LiPF}_6$  EC:DMC+2wt%LiTFSI and 1M  $\text{LiPF}_6$  SL:DMC +2wt%VC. Updated electrochemical data will be provided confirming the feasibility of LFMP/Graphite cells with increased energy density, nominal voltage 3.6 V, good power rate and stable cycle-life.

**Acknowledgement:** The SPICY project has received funding from the European Union's Horizon 2020 research and innovation programme (grant agreement n° 65373).