

Electrochemical properties of CoV_2O_4 : A novel anode material for sodium-ion batteries

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Transition metal oxides have been greatly studied on LIBs due to their high theoretical capacities ($>600 \text{ mAh}\cdot\text{g}^{-1}$) which is provided through conversion reaction with Li^+ [1, 2]. In LIB systems, numerous vanadium containing compounds have been studied such as, ZnV_2O_4 [3], CoV_2O_6 [4], CuV_2O_6 [5], FeVO_4 [6], and so forth. There are vanadium containing compounds that have already been applied to SIBs and showed promising results. This study reports the spinel based transition metal oxide – CoV_2O_4 (CVO) was synthesized by solvothermal method and annealed at 500°C . Its electrochemical performance as anode material for sodium ion battery was investigated for the first time. Ex-situ analyses were performed in order to understand the reaction mechanism that transpire during the charge-discharge cycle. The results showed that the initial reversible capacity of CVO was $115 \text{ mAh}\cdot\text{g}^{-1}$ at the current density of $100 \text{ mA}\cdot\text{g}^{-1}$ in the potential window of 0.01-3.0 V. The results indicate that CVO is a potential anode material for sodium ion battery.

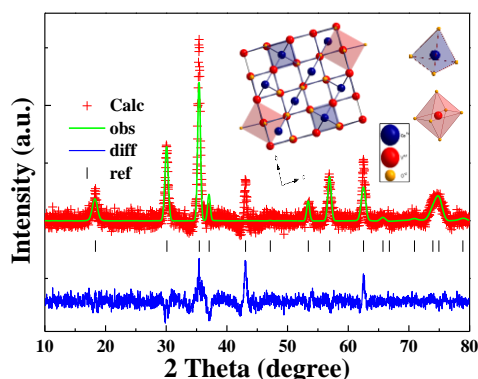


Fig. 1 Neutron powder diffraction data of CVO; observed, calculated (Rietveld) and difference intensity profiles and crystal structure.

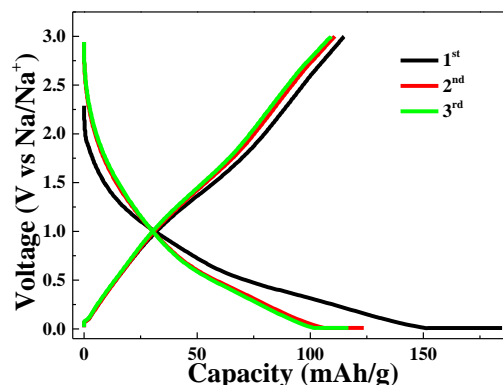


Fig. 2 Charge/discharge curves of CVO at $100 \text{ mA}\cdot\text{g}^{-1}$ for the first three cycles.

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