

## Selenium-doped graphene nanoplatelets for energy storage

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In this study, we investigated the effects of selenium doping in reduced graphene oxide (rGO) and energy storage performances of the selenium-doped rGO (Se-rGO). Numerous selenium atoms were homogeneously doped on the top surface of rGOs; it caused little structural change, indicative of surface transfer doping. High-resolution TEM images showed no aggregates or nanoparticles on the surface of rGOs, indicating that selenium atoms were introduced in an atomic level. Based on first principles calculations, we found the selenium atoms to be mainly bonded to the surface edge defect sites of highly amorphized rGO, leading to the enhancement of electrical conductivity due to n-type doping. In contrast, the overall intrinsic topological defects of the graphene basal plane showed an unfavorable selenium atom binding energy. The doped selenium atoms changed the electrical transport properties of rGO. The properties were well fitted by the fluctuation-induced tunneling mechanism, and the results indicated the presence of a large metallic region separated by a relatively small insulating gap. The  $\sigma$  value of Se-rGO reached  $210 \text{ S cm}^{-1}$  at 300 K, while the  $\sigma/\sigma_{20 \text{ K}}$  parameter was almost constant (1.3-fold increase from 20 to 300 K), this indicated that the carrier mobility of Se-rGO was similar to that of graphene. In addition, Se-rGO showed a superior electrochemical performance as an anode for Li-ion batteries, exhibiting  $\sim 570 \text{ mAh g}^{-1}$  at a current density of 1 C ( $372 \text{ mA g}^{-1}$ ). In a low-temperature cell test, the reversible capacity of Se-rGO is  $\sim 360 \text{ mAh g}^{-1}$ , corresponding to  $\sim 63\%$  of the capacity at  $25 \text{ }^\circ\text{C}$ . In contrast, rGO shows a reversible capacity of  $\sim 140 \text{ mAh g}^{-1}$ , corresponding to  $\sim 45\%$  of that at  $25 \text{ }^\circ\text{C}$ . These results suggest that the enhanced electrical properties of Se-rGO affect its electrochemical performance at low temperature to a great extent. Moreover, highly stable capacities of  $165 \text{ mAh g}^{-1}$  could be obtained at a large current rate of 50 C, which is approximately seven times higher than the value of rGO anode materials.