

Electrochemical and Electrocatalytic Behavior of Phosphate Class of Sodium Insertion Materials: Few Case Studies

Prabeer Barpanda, Ritambhara Gond, Lalit Sharma, Debasmita Dwibedi,
Sada Krishnakanth, Chinnasamy Murugesan, Baskar Senthilkumar
*Faraday Materials Laboratory, Materials Research Center, Indian Institute of Science,
C.V. Raman Avenue, Bangalore- 560012, India*

E-mail: prabeer@iisc.ac.in

Exploring post Li-ion battery systems, various monovalent (Na^+ , K^+) and multivalent (Mg^{2+} , Ca^{2+} , Zn^{2+} , Al^{3+}) chemistry have been intensely investigated over the last decade. Among them, sodium-ion based insertion compounds have emerged as frontrunners owing to their operational similarity with Li-ion systems, low cost/ abundance and high rate kinetics [1]. Various oxides and oxyanionic compounds have been reported with great success delivering high capacity and/or redox potential leading to promising energy density. In addition, some of these systems can be exploited as electrocatalysts with bi-functional activity.

Over the past two years, our group has investigated various phosphate (PO_4^{3-}) based insertion compounds for electrochemical and electrocatalytic activity [2-9]. In the current work, we will summarize these work to give some insights on sodium intercalation properties and bifunctional (oxygen evolution/ reduction reaction) electrocatalytic activity of following PO_4 materials.

(i) Solution combustion synthesis was used to prepare phase-pure $\text{Na}_2\text{FePO}_4\text{F}$ fluorophosphate compound involving low cost Fe(III) precursor. The target phase was obtained by annealing the intermediate complex in short duration of 1 minute. The electrochemical performance (3 V vs Na, 100 mAh/g) and one dimensional Na^+ diffusional mechanism will be demonstrated.

(ii) The electrochemical and diffusional activity of novel sodium metaphosphate [$\text{NaM}(\text{PO}_3)_3$] class of cathodes will be shown with a 2.8 V and 3.2 V (vs. Na) activity for Fe- and Co-based metaphosphate respectively. The electrocatalytic (oxygen reduction reaction) behavior of $\text{NaM}(\text{PO}_3)_3$ compounds will be demonstrated in comparison to Pt/C system.

(iii) The bifunctional activity of various phosphate (PO_4^{3-}) and pyrophosphate ($\text{P}_2\text{O}_7^{2-}$) materials (e.g. NaFePO_4 , KFePO_4 , NaCoPO_4 , $\text{Na}_2\text{CoP}_2\text{O}_7$, $\text{K}_2\text{CoP}_2\text{O}_7$) will be described and compared to Pt/C system. Most of these materials were found to be bifunctional in nature with potential application in Na-air batteries.

(iv) Finally, the electrochemical and electrocatalytic performance of phosphate based alluaudite materials [$\text{NaMFe}_2(\text{PO}_4)_3$] will be reported. An overall 3 V Na (de)intercalation was noticed with reversible capacity over 70 mAh/g along with efficient oxygen reduction reaction activity.

References:

- [1] N. Yabuuchi et al, S. Komaba, Chem. Rev. 114 (2014) 11636–11682.
- [2] L. Sharma et al, P. Barpanda, ACS Appl. Mater. Interfaces. 9 (2017) 34961–34969.
- [3] L. Sharma et al, P. Barpanda, Ionics. 24 (2018) In press.
- [4] C. Murugesan et al, P. Barpanda, ChemCatChem. 10 (2018) In press.
- [5] R. Gond et al, P. Barpanda, ChemElectroChem. 5 (2018) In press.
- [6] R. Gond et al, P. Barpanda, Inorg. Chem. 56 (2017) 5918–5929.
- [7] D. Dwibedi et al, P. Barpanda, ECS Trans. 80 (2017) 337–342.
- [8] D. Dwibedi et al, P. Barpanda, MRS Adv. (2018) In press.
- [9] D. Dwibedi et al, P. Barpanda, Electrochim. Acta. (2018) Submitted.